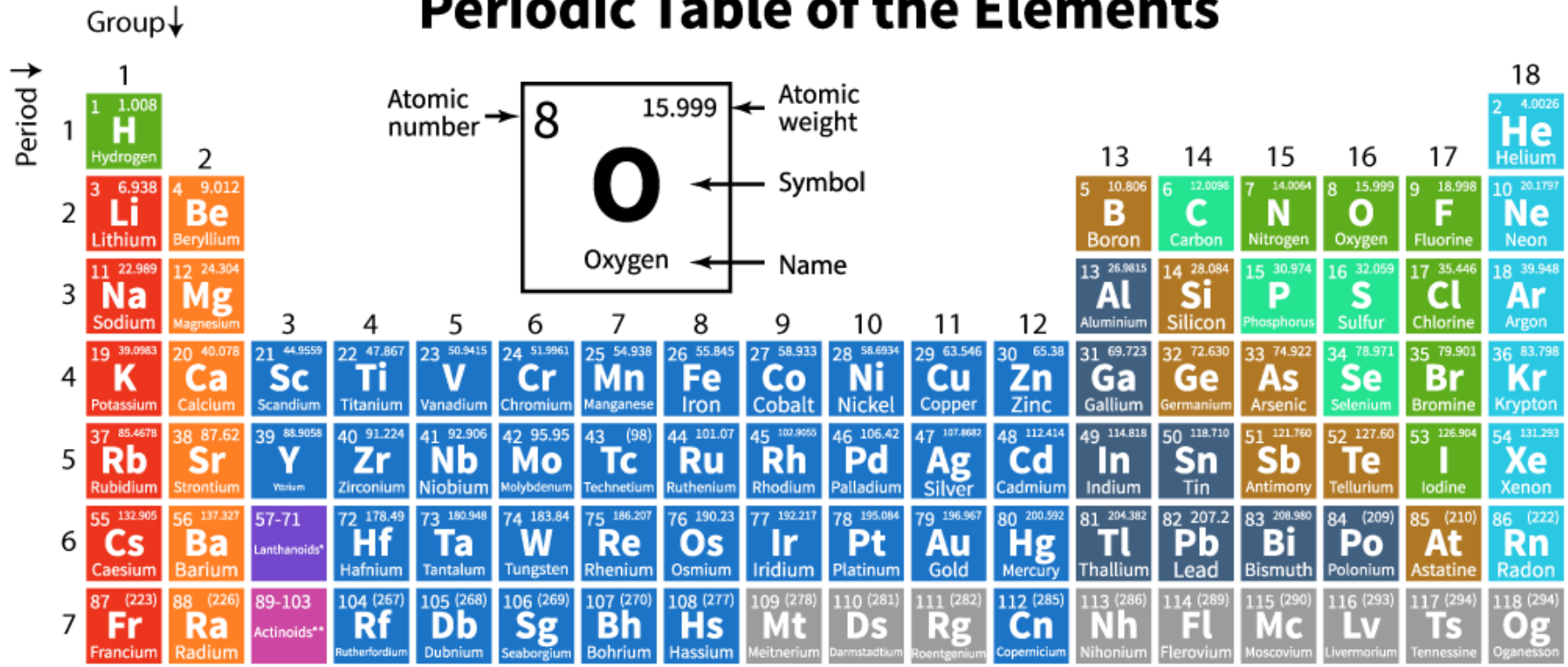
Неожиданные открытия

Интегрированный урок химии и английского языка

8 класс

Работу подготовили:
Кан Людмила Ивановна, учитель английского языка
Авдеева Оксана Сергеевна, учитель химии

Periodic Table of the Elements



Alkali Metals

Alkaline Earth Metals

Noble gas

Diatomic nonmetal

Unknown Chemical Properties

Lanthanide

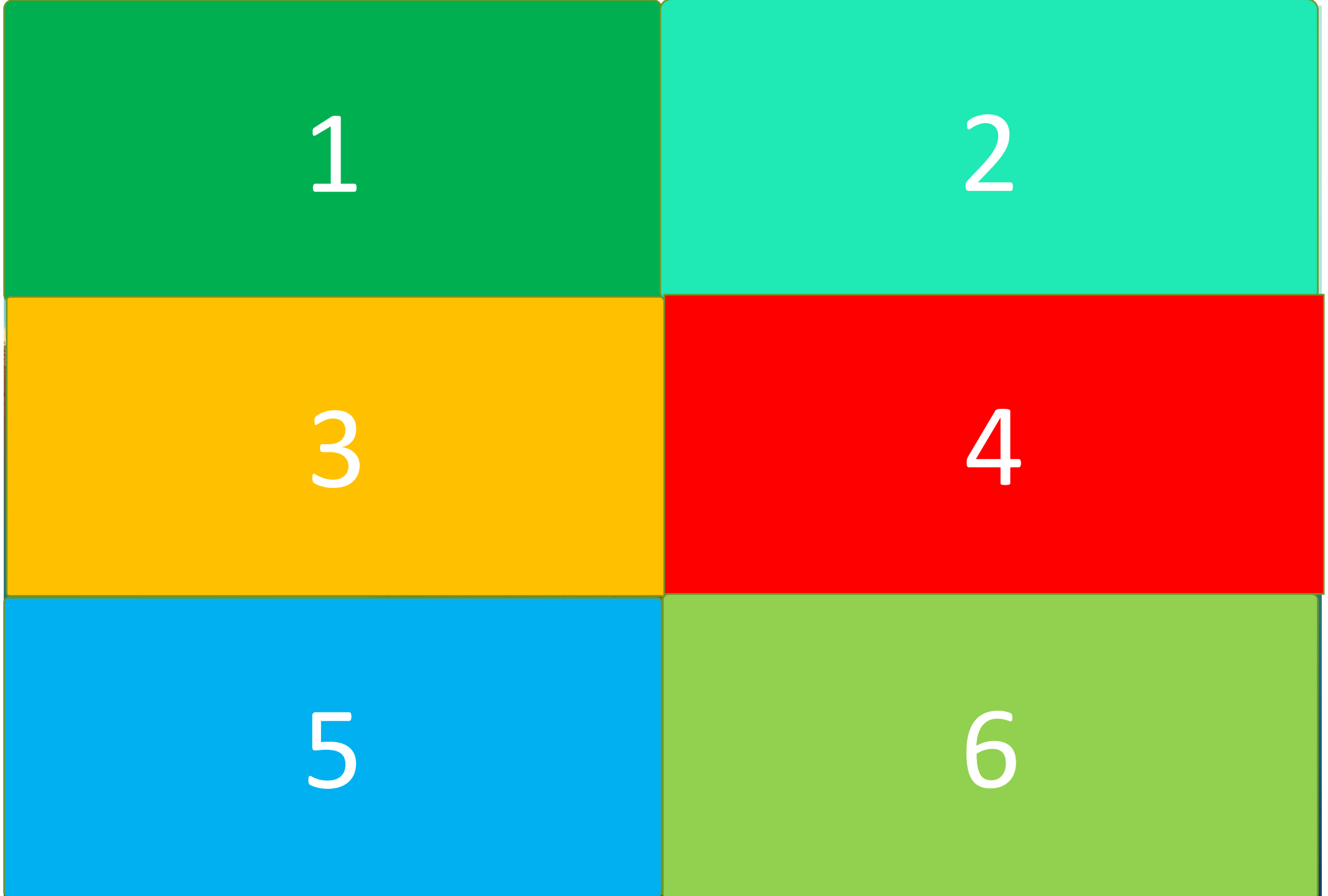
Actinide

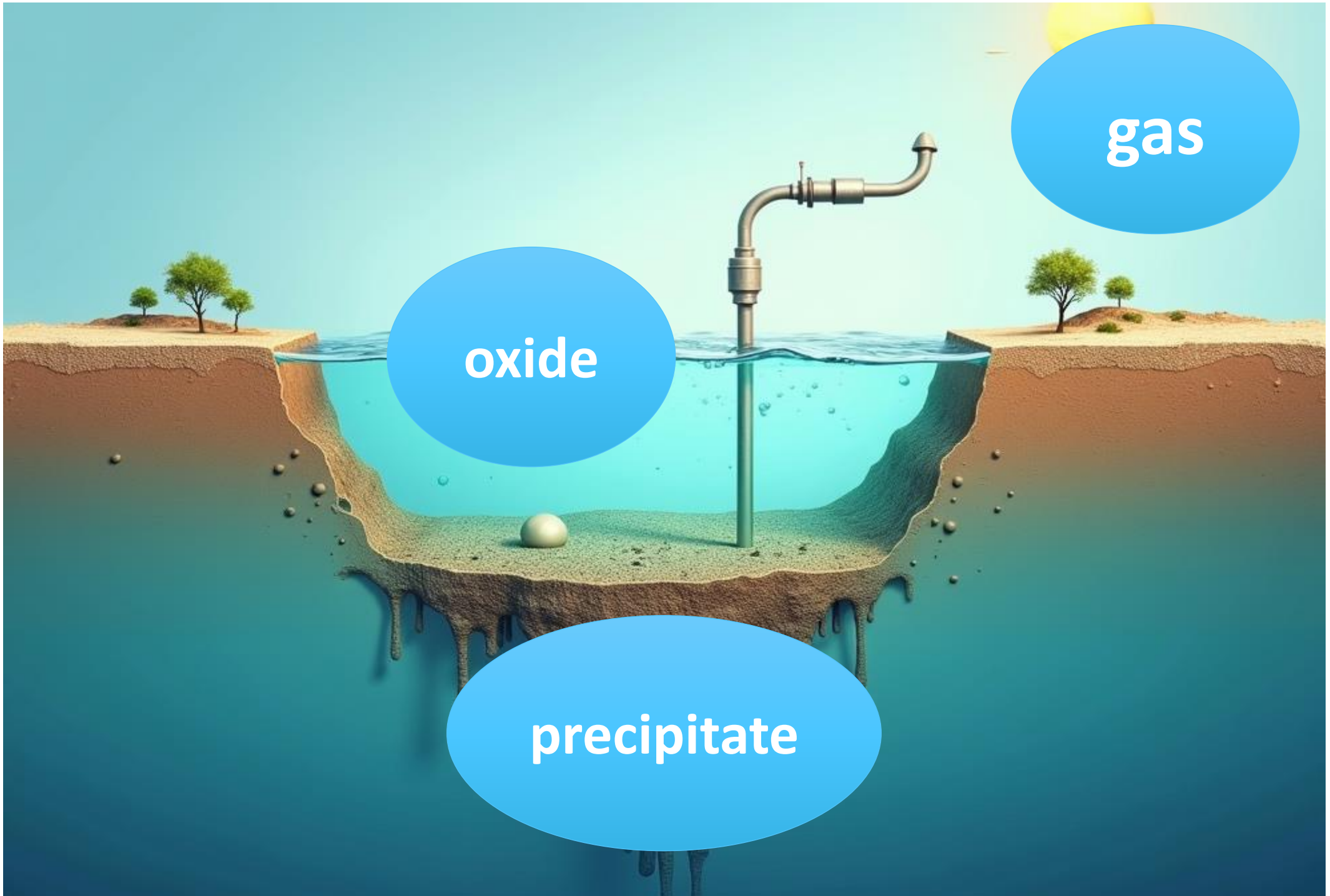
Transition Metals

Post-Transition Metals

Metalloid

Polyatomic nonmetal


1
2
3
4
5
6

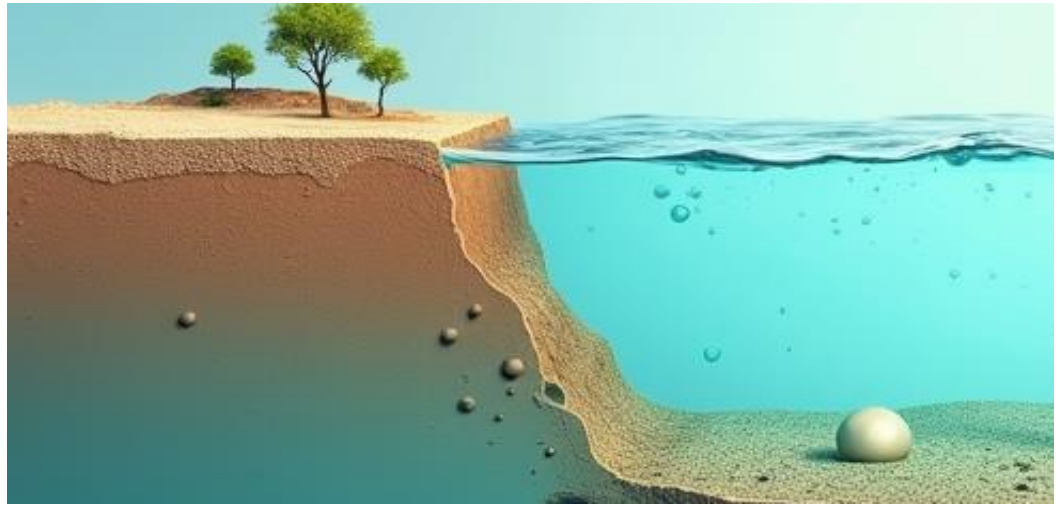

gas
oxide
precipitate

1

This substance is often used in the fire extinguishers. Atomic mass of 2 moleculs is 28.

2

It looks like a yellow solid substance at the ambient temperature compounded with preassure.



3

Its name was born from the Latin «natrium», which means «soda»

It is the most popular gas in the Earth’s crust and comprises 47% of the Earth’s crust.

4

5

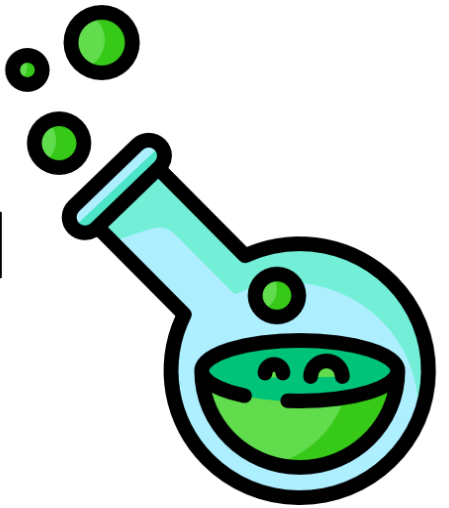
It absords X-radiation and gamma rays well enough. That’s why it is included in protaction materials. Its number is 56.

6

The name of this element comes from German word «zinke», which means «spiny»

Кислоты (acids)

H_2SO_4	sulfur**ic** acid
HNO_3	nitr**ic** acid
HCl	hydro-chlor**ic** acid
H_2CO_3	carbon**ic** acid
H_2S	hydrogen sulfide acid

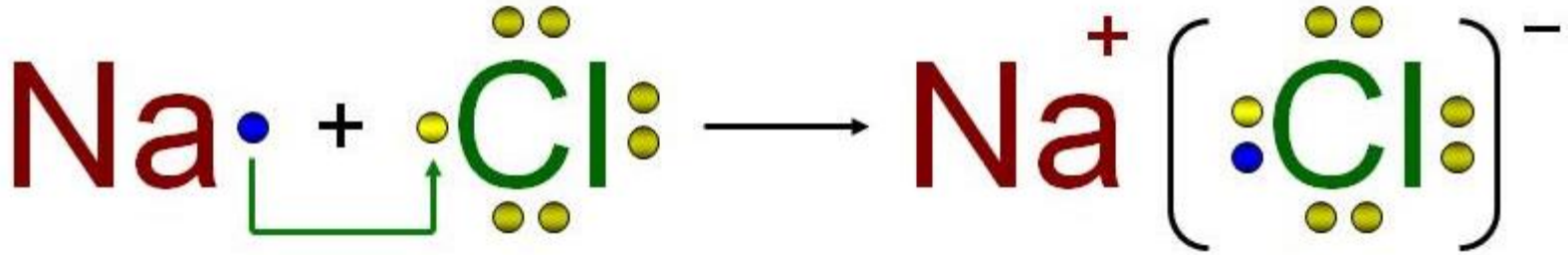


Work in groups

«X molecules and Y molecules react to form Z molecules»

$$2Na+Cl_2 = 2NaCl$$

(Two sodium molecules and chlorine react to form two sodium chloride molecules).



nitrate + sulfuric acid

H_2SO_4 NaOH HNO_3 KI HCl $CuCl_2$ H_2CO_3 Na_2S H_2S
$Ba(NO_3)_2$ $CuCl_2$ $CaCO_3$ $AgNO_3$ Zn

chloride + alkali

H_2SO_4 NaOH HNO_3 KI HCl $CuCl_2$ H_2CO_3 Na_2S H_2S
$Ba(NO_3)_2$ $CuCl_2$ $CaCO_3$ $AgNO_3$ Zn

hydro-chloric acid + carbonate

H_2SO_4 NaOH HNO_3 KI HCl $CuCl_2$ H_2CO_3 Na_2S H_2S
$Ba(NO_3)_2$ $CuCl_2$ $CaCO_3$ $AgNO_3$ Zn

nitrate + iodide

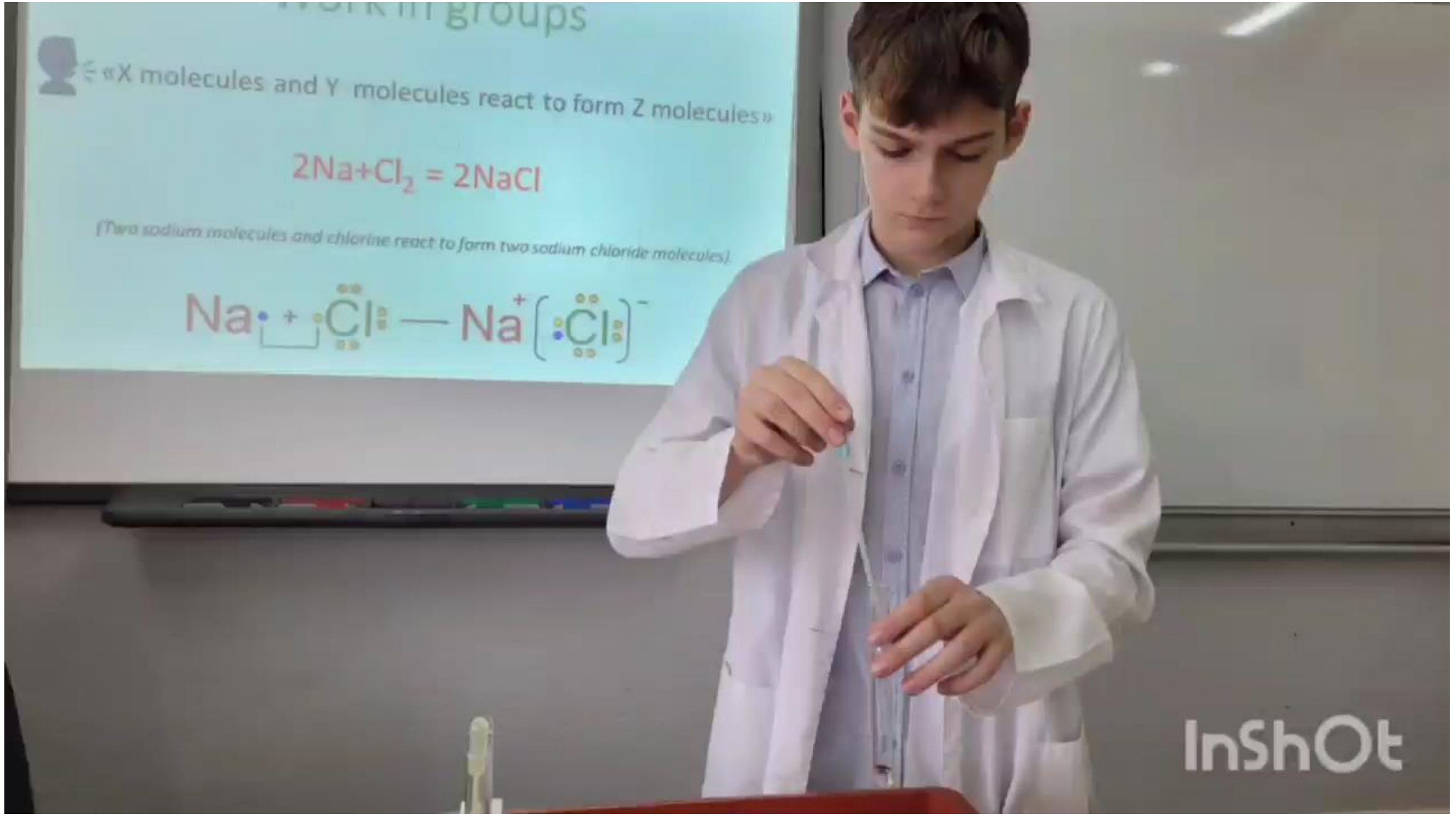
H_2SO_4 NaOH HNO_3 KI HCl $CuCl_2$ H_2CO_3 Na_2S H_2S
$Ba(NO_3)_2$ $CuCl_2$ $CaCO_3$ $AgNO_3$ Zn

hydro-chloric acid + zinc

H_2SO_4 NaOH HNO_3 KI HCl $CuCl_2$ H_2CO_3 Na_2S H_2S
$Ba(NO_3)_2$ $CuCl_2$ $CaCO_3$ $AgNO_3$ Zn

chloride + sulfide

H_2SO_4 NaOH HNO_3 KI HCl $CuCl_2$ H_2CO_3 Na_2S H_2S
$Ba(NO_3)_2$ $CuCl_2$ $CaCO_3$ $AgNO_3$ Zn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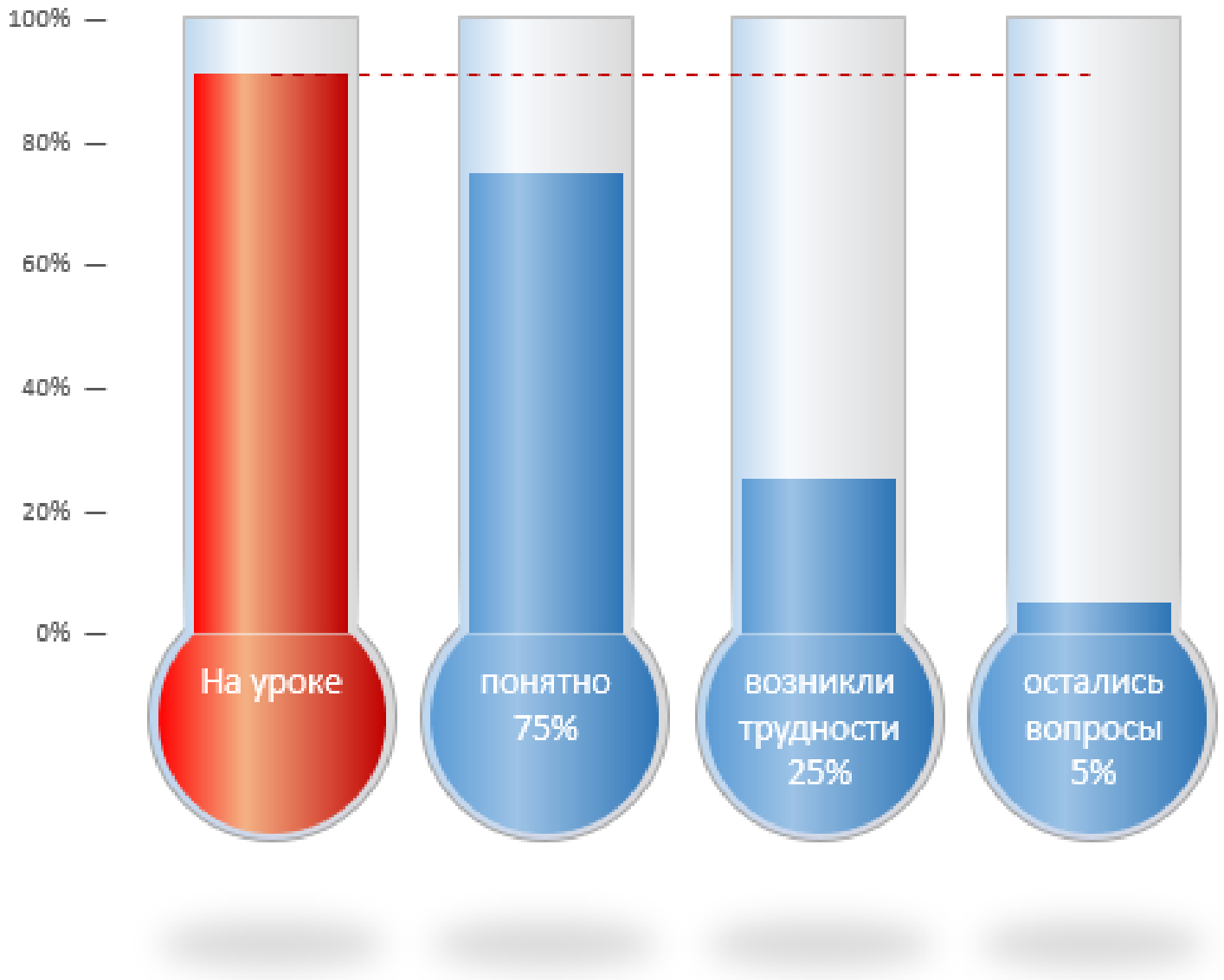
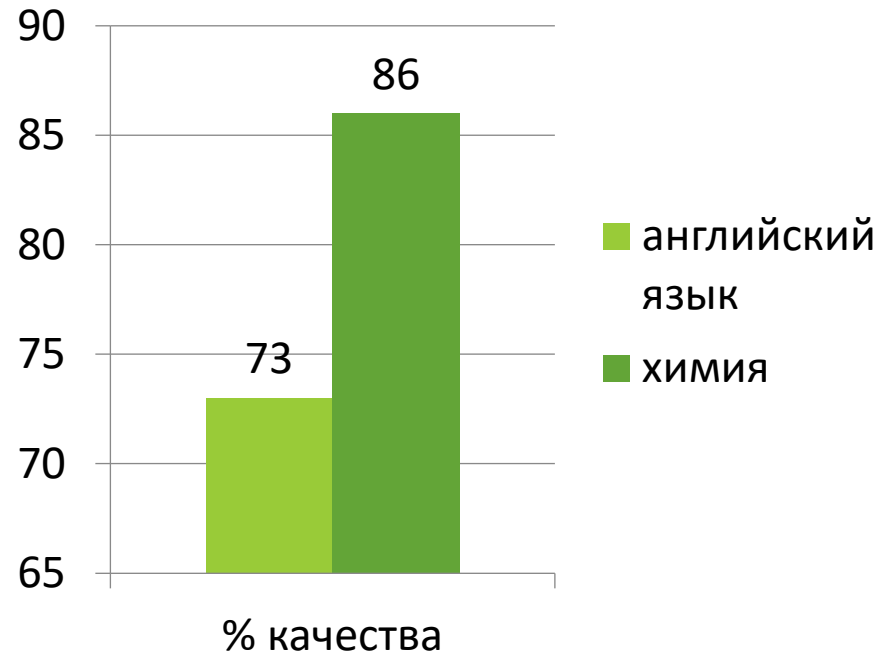
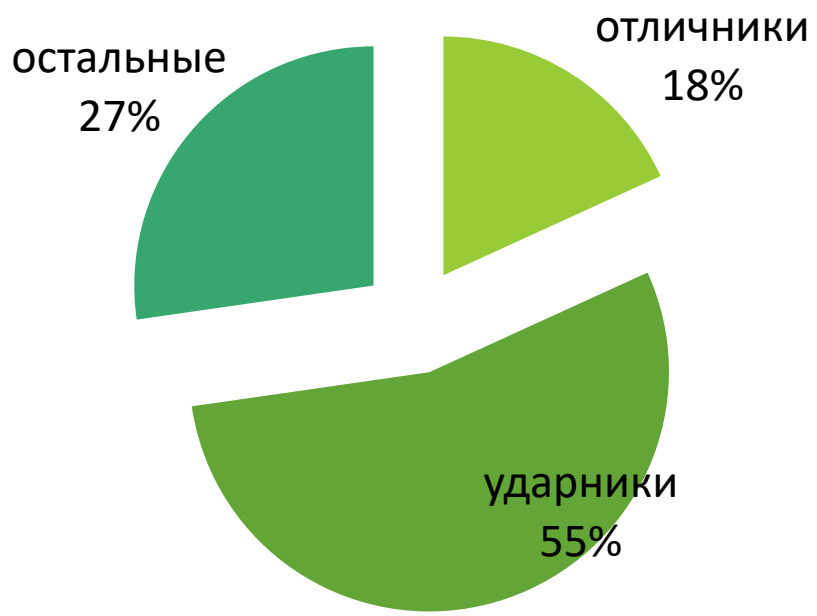
InShOt

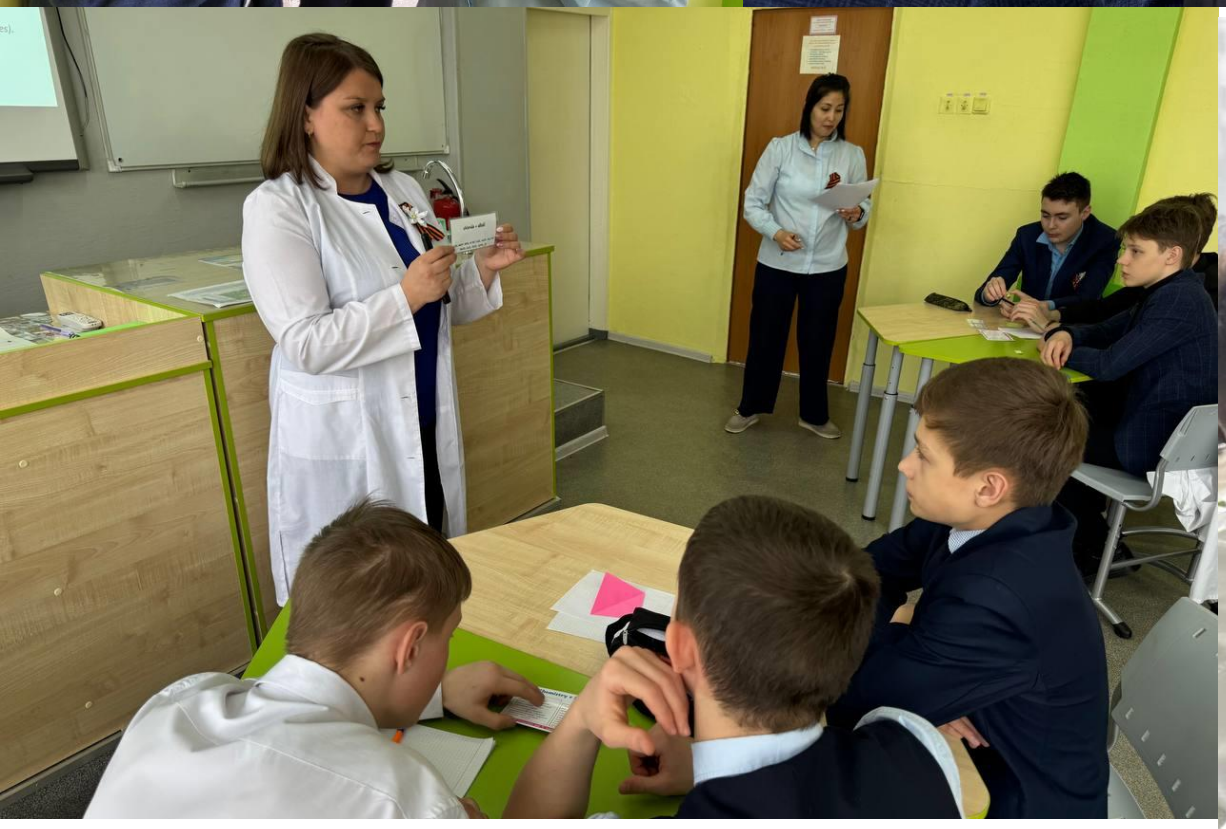

100%
80%
60%
40%
20%
0%
На уроке
понятно
75%
возникли
трудности
25%
остались
вопросы
5%


90
85
80
75
70
65
86
73
английский язык
химия
% качества


остальные 27%
отличники 18%
ударники 55%


ХИМИЕЙ


Chemistry + English

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